

<b>Study Programme:</b> Physics
<b>Course Unit Title:</b> Introduction to Condensed Matter Physics
<b>Course Unit Code:</b> F18OFKM
<b>Name of Lecturer(s):</b> dr Svetlana Lukić Petrović
<b>Type and Level of Studies:</b> Bachelor Academic Degree
<b>Course Status (compulsory/elective):</b> compulsory
<b>Semester (winter/summer):</b> winter
<b>Language of instruction:</b> English
<b>Mode of course unit delivery (face-to-face/distance learning):</b> face-to-face
<b>Number of ECTS Allocated:</b> 7
<b>Prerequisites:</b> Electromagnetism, Optics
<p><b>Course Aims:</b></p> <p>Acquiring basic knowledge of the internal structure of materials in the field of condensed matter physics and the study of interdependence in triad "synthesis-structure-properties"</p>
<p><b>Learning Outcomes:</b></p> <ul style="list-style-type: none"> <li>- Understanding the structure of ordered state, partially ordered state, nanostructure and amorphous materials.</li> <li>- Knowledge of methods for obtaining materials in condensed state and application possibilities</li> <li>- Knowledge of the specifics of certain types of materials as a result of dominant chemical bonds</li> <li>- Understanding the basic physical properties of solid materials</li> </ul>
<p><b>Syllabus:</b></p> <p><i>Theory</i></p> <p>Getting familiar with complex problems and concepts of condensed matter physics. Ordered state – crystals. Diffraction on crystal lattice. Defects in crystals.</p> <p>The nature of chemical bonds. Bond energy and parameters of phase transformations. Principles of structural ordering. Ionic crystals. Metals. Covalent and molecular crystals. Crystals with hydrogen bonding. Crystal complex.</p> <p>Getting familiar with relationship between structure and material properties; effect of processing on the structure and properties.</p> <p>Processes and technologies of obtaining condensed state materials.</p> <p>Liquid crystals, quasi crystals, polymers and nanostructures. Disordered systems. Phase diagrams and methods of obtaining amorphous materials.</p> <p>Condensed materials properties and methods of examination.</p> <p><i>Practice</i></p> <p>Experimental and computational exercises follow the content of lectures.</p>
<p><b>Required Reading:</b></p> <ol style="list-style-type: none"> <li>1. D.M. Petrović, S.R. Lukić, <i>Eksperimentalna fizika kondenzovane materije</i>, Edicija "Univerzitetski udžbenik", Univerzitet u Novom Sadu, Novi Sad, 2000</li> <li>2. Ch. Kittel, <i>Uvod u fiziku čvrstog stanja</i>, Savremena administracija, Beograd, 1970.</li> <li>3. H.M.Rosenberg, <i>The Solid State</i>, Oxford University Press, 1978.</li> <li>4. R.J.Elliott, A.F.Gibson, <i>Solid State Physics and its Applications</i>, Macmillan, Press Ltd., London, 1974.</li> <li>5. R.M.Rose, L.A.Shepard; <i>Struktura i osobine materijala</i>, Univerzitet u Novom Sadsu, Tehnološki fakultet, 2000</li> <li>6. P. Hofman, <i>Solid State Physics</i>, Wiley-VCH, New York, 2008.</li> <li>7. W. D. Callister, <i>Materials Science and Engineering: An Introduction</i>, John Wiley &amp; Sons, Inc., 2007.</li> <li>8. S. H. Simon, <i>The Oxford Solid State Basics</i>, Oxford University Press, Oxford, 2013</li> <li>9. J.K.Sirkin, M.E.Djatkina, <i>Hemijska veza i struktura molekula</i>, Građevinska knjiga, Beograd, 1957</li> </ol>

10. D. Grdenić, *Molekule i kristali*, Školska knjiga, Zagreb, 1979  
 11. Č. Jelačić, *Hemijska veza i struktura molekula*, Tehnička knjiga, Zagreb, 1982.  
 12. V. Šips, *Uvod u fiziku čvrstog stanja*, Školska knjiga, Zagreb..

<b>Weekly Contact Hours:</b>	<b>Lectures:3</b>	<b>Practical work:2</b>
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**Teaching Methods:**

**Knowledge Assessment (maximum of 100 points):**

<b>Pre-exam obligations</b>	points	<b>Final exam</b>	points
Active class participation	5	written exam	30
Practical work		oral exam	40
Preliminary exam(s)	15	.....	
Seminar(s)	10		

The methods of knowledge assessment may differ; the table presents only some of the options: written exam, oral exam, project presentation, seminars, etc.