Level: Bachelor
Course title: Calculation in Chemistry
Status: Elective
ECTS: 5
Requirements: None
Learning objectives

- Providing wide and balanced theoretical knowledge on key concepts of calculations in chemistry.
- Enabling students to apply standard methodology in solving calculation problems in chemistry.
- Providing the knowledge basis of fundamental chemical calculation for successful processing of knowledge in further chemical education.


## Learning outcomes

After successful completion of the course, a student is able to:

- Demonstrate the ability of abstract thinking on chemical calculation problems based on understanding of the fundamental chemical terms and definitions.
- Demonstrate knowledge and understanding of the basic concepts, terms and principles of homogenous and heterogeneous equilibria in water solutions.
- Practically apply theoretical knowledge and understanding in solving qualitative and quantitative problems.
- Recognize and solve chemical problems in familiar context and apply the acquired knowledge to other disciplines.


## Syllabus

## Theoretical instructions

Solutions. Mass fraction, concentrations, molality. Dilution and mixing of two solutions. Equilibria in water solutions: strong and weak electrolytes. Calculation of pH of strong acids and bases, weak acids and bases, polyprotic acids, buffers and ampholytes. Heterogeneous equilibria: solubility product, solubility. Precipitation. Influence of common ion. Quantitative and fractional precipitation.

## Practical instructions

Calculation of concentrations, pH in different solutions, and problems based on heterogeneous equilibria in water solutions.

| Weekly teaching load | Other: |  |  |  |
| :--- | :--- | :--- | :--- | :--- |
| Lectures: <br> 1 | Exercises: <br> 2 | Other forms of <br> teaching: <br> $/$ | Student research: <br> $/$ | $/$ |

