Level: Specialist academic studies of chemistry

Course title: Ionic Liquids (SH-612)

### Status: Elective

#### **ECTS**: 5

#### Requirements: None

#### Learning objectives

- Expanding knowledge and understanding of ionic liquids, one of the most modern and newest group of the compounds in chemistry and their application in industrial processes and laboratory conditions.
- Introducing students to interaction in the systems containing ionic liquids.

# Learning outcomes

Students should be able to:

- explain significance and importance of ionic liquids as a green solvents,
- list and explain interactions in ionic liquids media,
- explain the impact of some physical parameters (temperature, pressure, etc.) and individual components on physico-chemical characteristics of ionic liquids,
- demonstrate the need for further professional development.

# Syllabus

### *Theoretical instructions*

Structure and nomenclature of ionic liquids. Synthesis and purification of ionic liquids. Physicochemical characteristics of ionic liquids. Practical aspects and application of ionic liquids in green chemistry processes: liquid-liquid extraction, extraction of metal ions from different samples, application in gas chromatography, electrochemistry (sensors, biosensors), capillary electrophoresis. Ionic liquids as designed and green solvents. Ionic liquids in synthesis of organic and pharmaceutical compounds. Ionic liquids as electrolytes for energy storage.

# Practical instructions

The experiments are designed to illustrate the concepts discussed during the lectures and to familiarize students with ionic liquids. The lab is considered an integral part of the course.

Weekly teaching load				Other: /
Lectures:	Exercises:	Other forms of	Student research: /	
2	/	teaching: 2		