Level: Bachelor

**Course title:** Environmental chemistry

**Status**: obligatory

**ECTS**: 7

Requirements: none Learning objectives

To provide students with systematic knowledge of the most important areas of physical chemistry necessary for understanding the physical and chemical processes in the environment.

## **Learning outcomes**

General knowledge and understanding of the state of the matter, energy changes in physical and chemical processes, chemical and physical equilibrium, the processes at interfaces, chemical kinetics and photochemistry.

## **Syllabus**

Theoretical instruction

Liquid state. Surface tension and viscosity. Gaseous state. Ideal and real gas state. The laws of thermodynamics, thermochemistry, chemical potential. Chemical equilibrium. Thermodynamic principles of multi-component equilibria. Ideal and dilute solutions. Real solutions. Processes at interfaces. Air-water interface. Soil-water interface. Soil-air interface. Chemical kinetics. Fundamentals of colloidal chemistry. Photochemical processes.

## Practical instruction

Computational tasks from the major topics in the curriculum.

| Weekly teaching load |            |                |                     | Other: - |
|----------------------|------------|----------------|---------------------|----------|
| Lectures:            | Exercises: | Other forms of | Student research: - |          |
| 3 (45)               | 3 (45)     | teaching: -    |                     |          |